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Add Approximately 10 Drops Of Starch Indicator. A Blue-black Colour Should Be Seen As The Starch Reacts With The Residual Iodine. Continue To Add FA 1 1 cm^3 At A Time Until The Blue-black Colour Of The Starch-iodine Complex Disappears And Apr 3th, 2022

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CHEMISTRY 9701/01 1 Hour

10 When Vanadium(II) Compounds Are Dissolved In
Water, The Following Equilibrium Is Established. $V^{2+} + H_2O \rightleftharpoons V^{3+} + H_2O + OH^-$ What Would Alter The
Composition Of The Equilibrium Mixture In Favour Of
The V^{2+} Ions? A Adding An Acid B Adding A Reagent
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Fahad H. Ahmad (whatsapp: +92 323 509 4443)
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(c) (i) Write A Chemical Equation Representing The
Lattice Energy Of AgBr. (ii) Use The Following Data To
Calculate A Value For The Lattice Energy Of AgBr(s).
First Ionisation Energy Of Silver = +731kJmol⁻¹ Electron
Affinity Of Bromine = 325kJmol⁻¹ Enthalpy Change Of
Atomisation Of Silver = +285kJmol⁻¹ Enthalpy Change
Of Atomisation Of Bromine = +112kJmol⁻¹ ... Jan 10th,
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